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# **Structural Optimization and the Study of the Electronic, Mechanical, Thermodynamic and Phonon Properties of Mg<sub>2</sub>sn from First Principle**

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# *Authors' contributions*

*This work was carried out in collaboration among all authors. Author ISO proposed the article, provided all necessary resources for the acquisition of the data and training on how to use the software (Quantum Espresso) supervised and made the necessary corrections to the manuscript. Author TPC wrote the literature of the manuscript, performed the calculations with the aid of the aforementioned software and managed the analysis of the study. All authors read and approved the final manuscript.*

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# **ABSTRACT**

First principles pseudopotential method based on density functional theory is used to investigate the Structural, Mechanical, Phonon, Thermodynamic and Electronic properties of  $Mq_2Sn$ . The equilibrium properties including lattice constant, bulk modulus, pressure derivative cohesive energy, young modulus, shear modulus were determined. The results obtained were compared with available experimental and other available results. Mg<sub>2</sub>Sn was found to be brittle in nature with a non-metallic properties as shown by the value of the Cauchy pressure of -4.03. The Phonon dispersion curve of Mg2Sn was obtained utilizing the PBE-GGA exchange-correlation potential as employed in the Vienna Ab-Initio Simulation Package (VASP) computer code. The gap separating the acoustic and the optical branch of the curve was found to be about  $50 \text{cm}^{-1}$  at X-point. The

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thermodynamic properties of the material was investigated in the temperature of 0-800K. At room temperature, the calculated value of the specific heat capacity  $(C_v)$  is 71.28J/mol which is in good agreement with experimental and other results. Mg<sub>2</sub>Sn was found to a narrow gap semiconductor with an indirect bandgap of magnitude of 0.175eV.

*Keywords: Psuedopotential; phonon; density of state; thermodynamic; plane waves; pseudopotential; density functional theory.*

# **1. INTRODUCTION**

Magnesium is known to be the eight most common element in the Earth crust and the fourth most common element in the Earth (after iron, oxygen and silicon), making up to 13% of the planet's mass and large fraction of the planet's mantle.

Magnesium and its alloys have an excellent combination of properties of which some are excellent strength-to-weight ratio, good fatigue and impact strengths and relatively large thermal and electricalconductivities [1] and excellent biocompatibility [2]. Because of the above<br>mentioned properties, Magnesium has mentioned properties, Magnesium has applications in areas such as aerospace, Electronics, automotive and applications.

The stannide, germanide and silicide alloys of magnesium generally denoted as  $Mg_2X$  where X = Sn, Ge and Si, are a family of group II-IV compounds that are narrow gap semiconductors. These narrow gap semiconducting compounds show many advantages for potential TE (Thermoelectric) applications due to the abundance of constituents, low density, nontoxicity and environmental friendliness [3].

Among various Mg alloys, the Mg-Sn binary system is a typical precipitating, hardening system. It is a candidate for high strength wrought material due to high strength, excellent corrosion resistance, high temperature superplastic deformation and extrusion capability at moderate temperature. In addition,  $Mq_2$ Sn phase has a melting point of 1043K which resist dislocation slipping at high temperature and improves the mechanical properties at elevated temperature [4]. Apart from its promising thermoelectric properties,  $Mg<sub>2</sub>Sn$  compound has attracted interest as an electrode material for rechargeable lithium cells and for its catalytic properties.

The electronic and optical properties of  $Mg_2Sn$ have recently been studied by Uma Shankar Sharma [5], where the refractive index of  $Mg_2Sn$ wasfound to be 1.01. The work of Yu Rong Jin et [6] all has suggested that  $Mg_2$ Snis a promising mid- temperature thermoelectric material with maximum ZT value of 1.1 (for p-type  $Mg_2Sn$ ) with carrier concentration of  $9.8 \times 10^{19} \text{ cm}^3$ .

In this present work, the structural, electronic, mechanical, thermodynamic and phonon properties of Mg<sub>2</sub>Sn has been thoroughly investigated using quantum Espresso from first principle calculations and the values obtained are compared with other available studies.

# **2. METHODOLOGY AND COMPUTA TIONAL DETAILS**

The first principle calculation of this work was performed using Quantum Espresso package [7]. The exchange and correlation were described by Perdew-Burke-Erzerhof(PBE) functional in form of GGA [8]. PAW method was also used to generate the pseudopotential for the elements (Mg and Sn). The convergence of the total electronic energy as computed in plane wave pseudopotential code is determined by two important computational parameters, which are the number of basis functions (plane wave cutoff) and the number of K-points (k-spacing).

The number of basis functions was determined by running series of self-consistence calculations for different values of kinetic energy cut-off (ecutwfc) starting from fifty (50Ry) to hundred (100Ry) at an interval of 5Ry. The converged value of the ecutwfc is 80Ry. The value of the kpoints was also varied from a value of 4 kpointmesh to 16 k-point mesh at an interval of 2 k-point mesh. The two values help in determining accurately the electronic ground state properties of the system studied in the present work. A selfconsistent ab-initio calculation using the fullpotential linearized augmented plane wave (FP-LAPW) method within the framework of the spinpolarized density functional theory (DFT) was used to study the structural, electronic, mechanical, thermodynamic and phonon properties of Mg<sub>2</sub>Sn.

#### **3. RESULTS AND DISCUSSION**

#### **3.1 Structural Properties**

The antifluorite structure of  $Mg<sub>2</sub>Sn$  with space group Fm3m which has a faced centered cubic symmetry was studied in this present work. The crystal structure is shown in Fig. 1a below with the magnesium atoms in blue and tin atoms in ash colour.

The total energies as a function of lattice parameter are fitted to Murnaghan equation of state and the graphs of energy vs volume and pressure vs volume are shown in Fig.1. b-c, the equilibrium lattice constant (a), bulk modulus  $(B_0)$ and pressure derivative for  $Mg_2$ Sn were determined. The results obtained are shown in Table1. It can be seen that the result is in good agreement with available experimental and theoretical values.

#### **3.2 Mechanical Properties**

In order to get a full view of the mechanical properties of a material, the elastic constants of the material must be considered. Since  $Mg<sub>2</sub>Sn$ have a cubic structure, it is defined by three independent elastic constants which are  $C_{11}$ ,  $C_{12}$  and  $C_{44}$  and they are shown in Table 2 below. From Table 2, we can say that  $Mg_2Sn$  is mechanically stable because it satisfies the wellknown "Born stability criteria" for a cubic crystal [9].

$$
C_{11} - C_{12} > 0
$$
;  $C_{11} + 2C_{12} > 0$ ;  $C_{44} > 0$  (3.0.1)

Generally, the elastic constant  $C_{11}$  characterizes the x direction resistance to the linear compression,  $C_{44}$  is the most significant parameter, which indirectly determines the indentation hardness of a solid. A large  $C_{44}$ implies a strong resistance to monoclinic shear in the (1 0 0) plane.

Cauchy pressure,  $C_{12} - C_{44}$ , can be used to describe the angular character of atomic bonding in metals and compounds [10]. The negative value of the Cauchy-pressure implies that the material is non-metallic with directional bonding and if positive, the material is expected to be metallic. From Table 2, it is evident that the calculated value of the Cauchy-pressure of  $Mg<sub>2</sub>$ Sn at ambient conditions is negative, which indicates a non-metallic behavior of the material  $(i.eMg<sub>2</sub>Sn)$  is predicted to be a brittle material similarly as many non-metals).

The elastic modulus, including bulk modulus (B), shear modulus (G), Young's modulus (E), and Poisson's ratio  $(v)$  and also the B/G factor of Mg2Snadopting the Voigt-Reuss-Hill average approximation are presented in Table 3.

The correlation between the elastic properties of single crystals and an aggregate is made by employing the Voigt and Reuss continuum the ories [11]. Moreover, Hill demonstrated that the arithmetic average of the Voigt and Reuss values gives a better approximations for the elastic properties of materials. For cubic system, Voigt and Reuss bounds of B and G can be expressed as follows [12]:

$$
B_v = B_R = \frac{(C_{11} + 2C_{12})}{3}(3.0.2)
$$
  
\n
$$
G_v = \frac{(C_{11} - C_{12} + 3C_{44})}{3}(3.0.4)
$$
  
\n
$$
G_R = \frac{5(C_{11} - C_{12})C_{44}}{4(C_{44} + 3(C_{11} - C_{12})]}(3.0.5)
$$

And the arithmetic mean of B and G can be obtained as follow:

$$
B = \frac{1}{2}(B_V + B_R) \text{ and } G = \frac{1}{2}(G_R + G_V)(3.0.6)
$$

Young's modulus and Poisson's ratio are major elasticity related to characteristic property of a material, which are calculated using the following formula:

$$
E = \frac{9GB}{3B+G} \text{ and } v = \frac{3B-2G}{2(3B+G)}(3.0.7)
$$

From Table3 above, the calculated values for bulk modulus (B), young modulus (E), shear modulus (G) and Poisson ratio  $(v)$  are in close relation with experimental and other theoretical values.

The ratio of bulk modulus to shear modulus (B/G), proposed by Pugh [13], has been applied extensively to assess brittle or ductile behavior of materials. A high ratio, B/G, is associated with the ductile behavior of materials, whereas a low value corresponds to the brittle nature of materials. The critical value that separates ductile from brittle materials is 1.75. We see from the table above that  $Mg_2Sn$  is brittle in nature.

### **3.3 Phonon Properties**

An important branch of solid state physics is lattice dynamics, which concerns itself with the vibrations of atoms about their equilibrium sites in a solid. These vibrations are almost responsible for the thermal properties (like heat capacity, thermal conductivity, thermal expansion and so on) of a material [14].

In this present work, the calculation of phonon dispersion curve of Mg2Sn was performed utilizing the PBE-GGA exchange-correlation potential as employed in the Vienna Ab-Initio Simulation Package (VASP) computer code. The phonon dispersion curves are the aforemen tioned material is shown in Fig. 1d.

From Fig. 1d. there is a gap separating the acoustic branch and the optical branch of the curve. This gap was found to be about  $50 \text{cm}^{-1}$  at X-point. The states of lower energy (acoustic) are dominated by the tin and those of higher energy (optical) by the magnesium due to their important mass difference. The results obtained in this present work is in good relationship obtained by J bourgeois et al. [15].

# **3.4 Thermodynamic Properties**

The thermodynamic properties of  $Mg_2Sn$  which entails the specific capacity  $(C_n)$ , the Debye vibrational Energy, Debye free Energy, Debye entropy, the Debye temperature and sound velocity were successfully investigated in this present work. The calculations were performed in the temperature range of 0-800K.

Fig. 1e-1h below show the result of specific heat capacity at constant volume, the Debye entropy, Debye Free Energy, and Debye Vibrational Energy of Mg2Sn. From Fig. 1e. the specific heat capacity at constant volume increases rapidly as temperature increases and it approaches the Dulong-Petit limit at a very high temperature. At low temperature,  $C<sub>v</sub>$  is proportional to.<sup>73</sup>.

As temperature increases, the entropy of  $Mg_2Sn$ increases as seen in Fig. 1f. The free energy in Fig. 1g. below decreases as the temperature increases. From Fig. 1h, it is clear that the Debye vibrational energy increases as the temperature is increased. At room temperature (300K), the calculated value of the specific heat capacity  $(C_n)$ , Debye temperature  $(\theta_n)$  and the sound velocity of Mg<sub>2</sub>Sn are presented in Table 4.

### **3.5 Electronic Properties**

The Electronic band structure calculation for Mg2Sn was done using pseudo -potential and plane wave basis set method within the Density functional theory (DFT), treating exchangecorrelation functional with generalized gradient approximation (GGA) in the form of PredewBerke- Erzndof (PBE) functional as implemented in Quantum Espresso package.

Fig. 1i. and Fig. 1j, below show the band structure of Mg<sub>2</sub>Sn with the corresponding density of state (DOS).

From Fig. 1i, we see that the conduction band minimal state is at X-point which implies that  $Mg<sub>2</sub>Sn$  has an indirect bandgap. The calculated bandgap for  $Mg_2$ Sn is 0.175eV. The obtained result for the bandgap is in close agreement with 0.142eV obtained by Guangsha. S. and Emmanouil K. [16].

The DFT method in LDA/GGA is well known to underestimate the electronic band gap between materials [17]. An efficient method for the prediction of fundamental band gaps in solids using density functional theory (DFT) as proposed by M. K.Y. Chan and G. Ceder [18] is advisable.

The semiconducting character of this material can be determined from the total density of states (DOS) and the individual contribution of different orbitals in the partial density of states (PDOS). Fig. 1. j-k below show the total density of state (DOS) and the partial density of state (PDOS) of  $Mg_2$ Sn.

The density of state (DOS) provides numerical information on the states that are available at each energy level. The value of zero density of states indicates thatthere are no available states for occupation in an energetic level [19].

Detailed features of the total density of state (DOS) are shown by the partial density of state (PDOS) since it gives information about the origin of the bands. The fermi level  $(E_F)$  is set at zero point as shown in the Figs.  $1(j-k)$  below. The bands after (conduction band) the fermi energy are dense compared to that before the fermi energy (valence band). As a result of this, the conduction band contains more peaks. From Fig. 1k below, we see that we see that the Sn-3p state dominates the valence band while the Mg-2s state dominates the conduction.

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#### Table 2. Calculated elastic constants ( ${\cal C}_{11}, {\cal C}_{12}$   $and$   ${\cal C}_{44})$  and cauchy-pressure ( ${\cal C}$ **Mg2Sn**  ${\cal C}_{12}-{\cal C}_{44})$  of



*The values given in the above are in good agreement with experimental values and other recent works the other in* 

#### **Table 3. Voigt-reeuss-hill average of the two approximation of bulk modulus (B), young hill modulus (E), shear modulus (G) and poisson ratio ( )**



### Table 4. The specific heat capacity ( $C_v$ ), debye temperature ( $\theta_D$ ) and the sound velocity of **Mg2Sn**



*The values gotten for the heat capacity, Debye temperature and sound velocity are seen to be in The be the heat sound close agreement with available experimental and other results*



**Fig. 1c. Graph of raph pressure of Mg2Sn against volume**



**Fig. 1e. the specific heat capacity of Mg Mg2snin the temperature range of 0- -800K**



**Fig. 1f. The entropy of Mg he Mg2Snin the temperature range of 0-800K**



**Fig. 1g. Debye free energy of Mg2Sn in the temperature range of 0-800K 800K**



**Fig. 1h. Debye vibrational energy vibrational energy of Mg2Sn in the temperature range of 0- -800K**



**Fig. 1i. The electronic band structure of Mg2Sn**



**Fig. 1k. The partial density of state (PDOS) of Mg2Sn**

# **5. CONCLUSION**

The first principle calculation using Quantum Espresso package was successfully employed to optimize the structural properties and study the mechanical, phonon, thermodynamic and electronic properties of  $Mg<sub>2</sub>Sn$ .

The lattice parameter of the material was found to be 6.83Å which is in good agreement with available results.  $Mg_2$ Sn was found to be mechanically stable by satisfying the well-known "Born stability criteria". The B/G ratio of the material was found to be 1.55, which shows that Mg2Sn is brittle in nature. . The ground state properties and the elastic stiffness constants of this material are in good agreement with experimental and other results. From the phonon dispersion curve obtained in this work, the material is seen to be thermodynamically stable. The specific heat capacity and Debye temperature of  $Mg_2Sn$  was found to be 71.28J/mol and 310.5K which is in good agreement with experimental and other available results. From the electronic band structure, Mg<sub>2</sub>Sn was found to a narrow gap semiconductor with indirect bandgap of magnitude of 0.1751eV.

### **COMPETING INTERESTS**

Authors have declared that no competing interests exist.

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